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(54) Title: LIQUID BLEACHING COMPOSITION

(57) Abstract: The present invention relates to thickened pourable aqueous bleach composition which contains abrasives, suitable for use in cleaning hard surfaces such as are found, for example, in lavatories and kitchens which are highly alkaline.

### Liquid Bleaching Composition

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This invention relates to a pourable aqueous bleach composition which contains abrasives, suitable for use in cleaning hard surfaces such as are found, for example, in lavatories and kitchens.

It is particularly difficult to produce aqueous bleach stable, thickened abrasive cleaning compositions which offer the simultaneous benefits of: good bleach stability which provides an acceptable shelf life for a product; acceptable viscosity characteristics; relatively low cost; and of course efficacy in removing stains and in effectively disinfecting surfaces, especially on inclined hard surfaces such as tiled, enameled and porcelain surfaces such as are typically associated with bathrooms or kitchens. While the art provides a number of such compositions directed to satisfying these requirements, these compositions have not uniformly met with success. The present invention overcomes many of these shortcomings in the prior art.

A composition according to the invention is a stable, single phase, thickened bleach containing composition capable of adhering to vertical or inclined surfaces longer than thinner compositions. Such a composition is effective as an agent for stain and soil removal as well as disinfection. The high level of bleach stability and single solution phase behavior of the composition enables the composition to have an acceptable shelf life. The compositions exhibit little or no visibly discernible syneresis during its normal shelf life (6-9 months) and shelf life storage conditions (20°C.). In particularly preferred embodiments the compositions exhibit little or no syneresis under accelerated aging test conditions and at the same time exhibited good bleach stability. Such is particularly important for use and storage of the product in warm climate countries (South Africa, Brazil, etc.).

25

The present invention provides a liquid thickened chlorine bleach composition comprising an active chlorine source; at least one surfactant selected from anionic surfactants, amine oxides, and mixtures thereof;

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at least one thickener selected from polycarboxylates, polyacrylamides, clays, gums, cellulose derivatives, and mixtures thereof;

at least one abrasive material selected from oxides, carbonates, quartzes, siliceous chalk, diatomaceous earth, colloidal silicon dioxide, alkali metasilicates, organic abrasive materials selected from polyolefins, polyethylenes, polypropylenes, polyesters, polystyrenes, acetonitrile-

butadiene-styrene resins, melamines, polycarbonates, phenolic resins, epoxies and polyurethanes, natural materials selected from rice hulls, corn cobs, and the like, nepheline syenite, or talc and mixtures thereof;

water; and

5                   optionally, one or more materials selected from perfumes and perfume stabilizers, builders, rheology stabilizers; pH and buffering agents, electrolytes, pigments, colorants and the like.

Compositions according to the invention are alkaline in character, desirably having a pH of about 12 or more, exhibit very favorable viscosity characteristics, are good hard surface cleaners and feature 10 good bleach stability over time. The compositions may also include minor amounts, generally not more than at total of 10% wt., desirably less than 5% wt. of one or more optional constituents including ones which may improve the aesthetic appeal of the compositions, viz., perfumes and colorants. Such optional constituents should not undesirably affect the shelf stability or rheology of the compositions.

The chlorine source can be either organic or inorganic including alkali metal and alkaline earth 15 metal hypohalites, hypohalite addition products, haloamines, haloamines, haloimides, and haloamides. These also produce hypohalous bleaching species in situ. Preferred is hypochlorite and compounds producing hypochlorite in aqueous solution, although hypobromite is another potential halogen bleach. Those bleaching agents which yield a hypochlorite species in aqueous solution, include alkali metal and alkaline earth metal hypochlorites, hypochlorites addition products, chloramines, chloramines, 20 chloramides, and chlorimides. Examples of compounds of this type include sodium, potassium, lithium, and calcium hypochlorite, monobasic calcium hypochlorite, dibasic magnesium hypochlorite, chlorinated trisodium phosphate dodecahydrate, potassium dichloroisocyanurate, sodium dichloroisocyanurate, sodium dichloroisocyanurate dihydrate, trichlorocyanuric acid, 1,3-dichloro-5,5-dimethylhydantoin, N-chlorosulfamide, Chloramine T, Dichloramine T, Chloramine B and Dichloramine B.

25                   At lower pH levels, aqueous solutions formed by dissolving hypochlorite-yielding compounds contains active chlorine, partially in the form of hypochlorous acid moieties and partially in the form of hypochlorite ions. At pH levels above about 10, which is preferred for compositions containing hypochlorite, essentially all (greater than 99%) of the active chlorine is reported to be in the form of hypochlorite ion.

30                   Most of the above-described hypochlorite-yielding bleaching agents are available in solid or concentrated form and are dissolved in water during preparation of the compositions of the instant invention. Some of the above materials are available as aqueous solutions.

The above-described bleaching agents are dissolved in the aqueous liquid component of the present composition and are present in an active amount of from about 0.1% to 15% available chlorine.

Compositions of the present invention also include at least one surfactant selected from anionic surfactant, amine oxides, and mixtures thereof. By way of non-limiting example useful anionic surfactants include, for example, alkali metal salts, ammonium salts, amine salts, or aminoalcohol salts of one or more of the following compounds (linear and secondary): alcohol sulfates and sulfonates, alcohol phosphates and phosphonates, alkyl sulfates, alkyl ether sulfates, sulfate esters of an alkylphenoxy polyoxyethylene ethanol, alkyl monoglyceride sulfates, alkyl sulfonates, olefin sulfonates, paraffin sulfonates, beta-alkoxy alkane sulfonates, alkylamidoether sulfates, alkylaryl polyether sulfates, monoglyceride sulfates, alkyl ether sulfonates, ethoxylated alkyl sulfonates, alkylaryl sulfonates, alkyl benzene sulfonates, alkylamide sulfonates, alkyl monoglyceride sulfonates, alkyl carboxylates, alkyl sulfoacetates, alkyl ether carboxylates, alkyl alkoxy carboxylates having 1 to 5 moles of ethylene oxide, alkyl sulfosuccinates, alkyl ether sulfosuccinates, alkylamide sulfosuccinates, alkyl sulfosuccinamates, octoxynol or nonoxynol phosphates, alkyl phosphates, alkyl ether phosphates, taurates, N-acyl taurates, fatty taurides, fatty acid amide polyoxyethylene sulfates, isethionates, acyl isethionates, and sarcosinates, acyl sarcosinates, or mixtures thereof. Generally, the alkyl or acyl radical in these various compounds comprise a carbon chain containing 12 to 20 carbon atoms.

Examples of the foregoing anionic surfactants are available under the following tradenames: RHODAPON, STEPANOL, HOSTAPUR, SURFINE, SANDOPAN, NEODOX, BIOSOFT, and AVANEL.

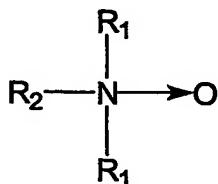
Amine oxide compounds which can be used in the present invention may be defined as one or more of the following of the four general classes:

- (1) Alkyl di (lower alkyl) amine oxides in which the alkyl group has about 6-24, and preferably 8-18 carbon atoms, and can be straight or branched chain, saturated or unsaturated. The lower alkyl groups include between 1 and 7 carbon atoms, but preferably each include 1 – 3 carbon atoms. Examples include octyl dimethyl amine oxide, lauryl dimethyl amine oxide, myristyl dimethyl amine oxide, and those in which the alkyl group is a mixture of different amine oxides, such as dimethyl cocoamine oxide, dimethyl (hydrogenated tallow) amine oxide, and myristyl/palmityl dimethyl amine oxide;
- (2) Alkyl di (hydroxy lower alkyl) amine oxides in which the alkyl group has about 6-22, and preferably 8-18 carbon atoms, and can be straight or branched chain, saturated or unsaturated. Examples include bis-(2-hydroxyethyl) cocoamine oxide, bis-(2-hydroxyethyl) tallowamine oxide; and bis-(2-hydroxyethyl) stearylamine oxide;

(3) Alkylamidopropyl di(lower alkyl) amine oxides in which the alkyl group has about 10-20, and preferably 12-16 carbon atoms, and can be straight or branched chain, saturated or unsaturated. Examples are cocoamidopropyl dimethyl amine oxide and tallowamidopropyl dimethyl amine oxide; and

5 (4) Alkylmorpholine oxides in which the alkyl group has about 10-20, and preferably 12-16 carbon atoms, and can be straight or branched chain, saturated or unsaturated.

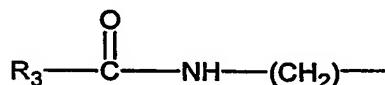
While these amine oxides recited above may be used, preferred are amine oxides which may be represented by the following structural representation:



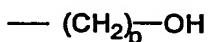
wherein

.0 each  $R_1$  independently is a straight chained  $C_1-C_4$  alkyl group; and,

$R_2$  is a straight chained  $C_6-C_{22}$  alkyl group or an alkylamidoalkylene having the formula



where  $R_3$  is  $C_5-C_{20}$  alkyl or



15 where  $n$  is 1 to 5 and  $p$  is 1 to 6; additionally,  $R_2$  or  $R_3$  could be ethoxylated (1 to 10 moles EO/mol) or propoxylated (1 to 10 moles of PO/mol).

Each of the alkyl groups may be linear or branched, but most preferably are linear. Examples include Ammonyx® LO which is described to be as a 30%wt. active solution of lauryl dimethyl amine oxide; Ammonyx® CDO Special, described to be a about 30%wt. active solution of cocoamidopropylamine oxide, as well as Ammonyx® MO, described to be a 30%wt. active solution of myristyldimethylamine oxide, all available from Stepan Company (Northfield, IL) with similar materials 20 also available from Lonza under the Barlox trademark.

Some of the aforementioned surfactants are bleach-stable but some are not. When the composition contains a hypochlorite bleach, it is preferable that the detergent surfactant is bleach-stable. Such surfactants desirably do not contain functions such as unsaturation and some aromatic, amide, 25 aldehydic, methyl keto or hydroxyl groups which are susceptible to oxidation by the hypochlorite.

The surfactant component, selected from anionic surfactants, amine oxides, and mixtures thereof, is present in the compositions of the present invention in an amount of from about 0.1 to about 10% by weight.

5 At least one thickener is used in the present compositions. One example of thickeners contemplated herein include organic polymeric thickeners include polycarboxylate polymers having a molecular weight from about 500,000 to about 4,000,000, preferably from about 1,000,000 to about 4,000,000, with, preferably, from about 0.5% to about 4% crosslinking. Preferred polycarboxylate polymers include polyacrylate polymers including those sold under trade names Carbopol®, Acrysol®  
10 ICS-1 and Sokalan®. The preferred polymers are polyacrylates. Other monomers besides acrylic acid can be used to form these polymers including such monomers as ethylene and propylene which act as diluents, and maleic anhydride which acts as a source of additional carboxylic groups. Another example of polymeric based thickeners are those based on polyacrylamides. One example is Solagum from Sepic.

15 Another class of thickeners include colloid-forming clays, for example, such as smectite and/or attapulgite types. Inorganic colloid forming clays tend to provide higher stability in the presence of chlorine and do not thin when subjected to shear.

The clay materials can be described as expandable layered clays, i.e., aluminosilicates and magnesium silicates. The term "expandable" as used to describe the instant clays relates to the ability of  
20 the layered clay structure to be swollen, or expanded, on contact with water. The expandable clays used herein are those materials classified geologically as smectites (or montmorillonite) and attapulgites (or polygorskites). Smectites are three-layered clays. There are two distinct classes of smectite-type clays. In the first, aluminum oxide is present in the silicate crystal lattice; in the second class of smectites, magnesium oxide is present in the silicate crystal lattice. The general formulas of these smectites are  
25  $\text{Al}_2(\text{Si}_2\text{O}_5)_2(\text{OH})_2$  and  $\text{Mg}_3(\text{Si}_2\text{O}_5)(\text{OH})_2$ , for the aluminum and magnesium oxide type clays, respectively. It is to be recognized that the range of the water of hydration in the above formulas may vary with the processing to which the clay has been subjected.

Commercially available clays include, for example, montmorillonite (bentonite), volchonskoite, nontronite, beidellite, hectorite, saponite, sauconite and vermiculite. The clays herein are available under  
30 various trade names such as Gelwhite H NF and Gelwhite GP from Southern Clay Products. (both montmorillonites); Van Gel O from R. T. Vanderbilt, smectites, laponites and layered silicates from Southern Clay Products. A second type of expandable clay material useful in the instant invention is

classified geologically as attapulgite (polygorskite). Attapulgites are magnesium-rich clays having principles of superposition of tetrahedral and octahedral unit cell elements different from the smectites. Like the smectites, attapulgite clays are commercially available. For example, such clays are marketed under the tradename Attagel, i.e. Attagel 40, Attagel 50 and Attagel 150 from Engelhard Minerals &

5 Chemicals Corporation.

The thickener component may be used in amounts of about 0.1 to 10 wt-%.

Another component of the present invention is at least one abrasive. Examples of abrasive materials include oxides, carbonates, quartzes, siliceous chalk, diatomaceous earth, colloidal silicon dioxide, alkali metasilicates, organic abrasive materials selected from polyolefins, polyethylenes, 10 polypropylenes, polyesters, polystyrenes, acetonitrile-butadiene-styrene resins, melamines, polycarbonates, phenolic resins, epoxies and polyurethanes, natural materials selected from rice hulls, corn cobs, and the like, nepheline-syenite, or talc and mixtures thereof. The particle size of the abrasive agent can range from about 1  $\mu\text{m}$  to about 1000  $\mu\text{m}$ , preferably between about 10  $\mu\text{m}$  to about 200  $\mu\text{m}$ , and more preferably between about 10  $\mu\text{m}$  and about 100  $\mu\text{m}$ . It is preferred to us those abrasive agents 15 that will not scratch glass or ceramic surfaces. Such abrasive agents include calcium carbonate, siliceous chalk, diatomaceous earth, colloidal silicon dioxide, sodium metasilicate, talc, and organic abrasive materials. Calcium carbonate is preferred. The amount of abrasive material can range from about 5 to about 30 wt%.

As noted the compositions may also include minor amounts, generally not more than at total of 20 10% wt., desirably less than 5% wt. of one or more optional constituents including ones which may improve the aesthetic appeal of the compositions, viz., perfumes and colorants. Such optional constituents should not undesirably affect the shelf stability or rheology of the compositions. By way of non-limiting example such further constituents include one or more coloring agents, fragrances and fragrance 25 solubilizers, viscosity modifying agents, other surfactants, pH adjusting agents and pH buffers including organic and inorganic salts, optical brighteners, opacifying agents, hydrotropes, antifoaming agents, enzymes, anti-spotting agents, anti-oxidants, preservatives, and anti-corrosion agents. When one or more of the optional constituents is added, i.e., fragrance and/or coloring agents, the esthetic and consumer appeal of the product is often favorably improved. The use and selection of these optional constituents is well known to those of ordinary skill in the art.

30

#### Examples

For purposes of illustration of the present inventive compositions, various formulations were produced and are reported below.

5 Into a suitably sized vessel, a measured amount of room temperature water (~ 20°C) was provided after which measured amounts of remaining constituents were added. All of the remaining constituents were similarly supplied at room temperature; mixing of the constituents was achieved by the use of a magnetic stirrer apparatus. Mixing, which generally lasted several minutes, and maintained until the particular exemplary formulation appeared to be homogeneous.

It is to be noted that the constituents may be added in any order. In some instances, the thickener is added first and then followed by water to form a thickened system to which the other components are added and in other instances, a premix of water and thickener is made which is then added to another mixture which contains the remaining components.

10 Exemplary inventive formulations according to the present invention including certain specifically preferred embodiments are shown in Table 1. The quantities of composition components are given as percentages-by-weight (wt%). Unless otherwise noted, the components in Table 1 are at 100% active.

Table 1

Components	Ex. 1 Wt %	Ex. 2 Wt %	Ex. 3 Wt %	Ex. 4 Wt %	Ex. 5 Wt %	Ex. 6 Wt %
DI Water	41.28	39.33	36.33	39.28	30.00	33.00
Carbopol 676 (1.5%)	26.67	26.67	26.67	26.67	30.00	30.00
Oxy-Rite 100	0.05			0.05	13.00	
Gelwhite H NF (5%)	10.00	10.00	10.00	10.00		10.00
Sodium Hydroxide (50%)	1.50	1.50	1.50	1.50	1.50	1.50
Sodium Hypochlorite (16%)	7.00	7.00	7.00	7.00	7.00	7.00
Calcium Carbonate	10.00	12.00	15.00	12.00	15.00	15.00
Ammonyx MO	1.00	1.00	1.00	1.00	1.00	1.00
Genopur BK	2.50	2.50	2.50	2.50	2.50	2.50
Total	100.00	100.00	100.00	100.00	100.00	100.00

Table 1

Components	Ex. 7 Wt %	Ex. 8 Wt %	Ex. 9 Wt %	Ex. 10 Wt %	Ex. 11 Wt %	Ex. 12 Wt %
DI Water	23.50	42.00	43.00	39.50	43.00	33.33
Gelwhite GP (5%)	10.00	5.00	5.00	3.00	5.00	15.00
Carbopol 676 (1.5%)	42.00	33.00	30.00	35.00	33.00	26.67
Sodium Hydroxide (50%)	1.50	1.50	1.50	1.50	1.50	1.50
Sodium Hypochlorite	7.00	7.00	7.00	7.00	7.00	7.00
Calcium Carbonate	12.50	9.00	10.00	10.00	7.00	12.00
Titanium Dioxide				0.50	0.50	
Ammonyx MO	1.00	1.00	1.00	1.00	1.00	1.00
Genopur BK	2.50	1.50	2.50	2.50	2.00	3.50
Total	100.00	100.00	100.00	100.00	100.00	100.00

**Table 1**

	Ex. 13	Ex. 14	Ex. 15	Ex. 16	Ex. 17	Ex. 18
Components	Wt %					
DI Water	36.33	38.00	41.33	28.50	33.50	27.00
Carbopol 676 (1.5%)	26.67	30.00	26.67	45.00	40.00	30.00
Gelwhite H NF (5%)	15.00	10.00	10.00			16.00
Sodium Hydroxide (50%)	1.50	1.50	1.50	1.00	1.00	1.00
Sodium Lauryl Ether Sulfate				1.50	1.50	1.50
Sodium Carbonate						0.50
Sodium Silicate				1.00	1.00	
Sodium Hypochlorite (16%)	7.00	7.00	7.00	7.00	7.00	7.00
Calcium Carbonate	10.00	10.00	10.00	15.00	15.00	16.00
Ammonyx MO	1.00	1.00	1.00	1.00	1.00	1.00
Genopur BK	2.50	2.50	2.50			
<b>Total</b>	<b>100.00</b>	<b>100.00</b>	<b>100.00</b>	<b>100.00</b>	<b>100.00</b>	<b>100.00</b>

**Table 1**

	Ex. 19	Ex. 20	Ex. 21	Ex. 22	Ex. 23	Ex. 24
Components	Wt %					
DI Water	41.33	10.00	19.00	36.00	10.00	20.03
Carbopol 676 (1.5%)	26.67	30.00	54.00	35.00	10.00	26.67
Van Gel O					50.00	
Mineral Colloid BP						30.00
Laponite (4%)		37.50				
Gelwhite H NF (5%)	10.00		1.00	5.00		
Sodium Hydroxide (50%)	1.50	1.00	2.00	1.50	1.00	0.30
Sodium Carbonate		1.00			1.00	
Sodium Hypochlorite (16%)	7.00	7.00	7.00	7.00	7.00	7.00
Sodium Lauryl Ether Sulfate					1.00	1.00
Calcium Carbonate	10.00	12.00	15.00	12.00	20.00	15.00
Ammonyx MO	1.00			1.00		
Biosoft D-40		1.50	2.00			
Genopur BK	2.50			2.50		
<b>Total</b>	<b>100.00</b>	<b>100.00</b>	<b>100.00</b>	<b>100.00</b>	<b>100.00</b>	<b>100.00</b>

**Table 1**

	Ex. 25	Ex. 26	Ex. 27	Ex. 28	Ex. 29	Ex. 30
Components	Wt %					
DI Water	36.33	33.00	33.00	50.33	20.00	17.50
Gelwhite GP (5%)	15.00	15.00	15.00			
Solagum SF 306				1.00		
Laponite (5%)					28.00	30.00
Carbopol 676 (1.5%)	26.67	30.00	30.00	26.67	30.00	30.00
Sodium Hydroxide (50%)	1.50	1.50	1.50	1.50	1.50	1.50
Sodium Hypochlorite (16%)	7.00	7.00	7.00	7.00	7.00	7.00
Calcium Carbonate	10.00	10.00	10.00	10.00	10.00	10.00
Ammonyx MO	1.00	1.00	1.00	1.00		3.50
Genopur BK			2.50			0.50
Sodium Lauryl Ether Sulfate	2.50	2.50		2.50	3.50	
Total	100.00	100.00	100.00	100.00	100.00	100.00

The identity of the individual components are described in Table 2.

**Table 2**

DI Water	Deionized water
Carbopol 676 (1.5%)	Highly cross-linked polyacrylate polymer (Noveon; 100% active; added as 1.5% active solution in water)
Oxy-Rite 100	Aromatic based rheological stabilizer (Noveon; 100% active)
Gelwhite H NF (5%)	Montmorillonite (Southern Clay Products, Inc.; 100% active; added as 5% active solution in water)
Sodium Hydroxide (50%)	Sodium hydroxide (50% active solution in water)
Sodium Hypochlorite (16%)	Sodium hypochlorite (16% active solution in water)
Calcium Carbonate	Calcium carbonate
Ammonyx MO	Myristyldimethylamine oxide (Stepan; 30% active)
Genopur BK	Mixture of anionic surfactants (Clariant)
Gelwhite GP (5%)	Montmorillonite (Southern Clay Products, Inc.; 100% active; added as 5% active solution in water)
Titanium Dioxide	Titanium oxide
Sodium Lauryl Ether Sulfate	Sodium lauryl ether sulfate
Sodium Carbonate	Sodium carbonate
Sodium Silicate	Sodium silicate
Van Gel O	Hydrated magnesium aluminum silicate (smectite) clay (RT Vanderbilt)
Mineral Colloid BP	high purity montmorillonite (Southern Clay Products, Inc.; 100% active)
Laponite (4%); Laponite (5%)	Synthetic layered magnesium silicate (Southern Clay Products, Inc.; 100% active; added as a 4% (or 5%) active solution in water)
Biosoft D-40	Sodium linear alkylbenzene sulfonate (Stepan; 38% active)
Solagum SF 306	Polyacrylamide co-polymer (Seppic)

The compositions of the present invention are expected to have good cleaning and/or disinfecting properties as described herein.

## Claims:

1. A liquid thickened chlorine bleach composition comprising:
  - 5 an active chlorine source;
  - at least one surfactant selected from anionic surfactants, amine oxides, and mixtures thereof;
  - 10 at least one thickener selected from polycarboxylates, polyacrylamides, clays, gums, cellulose derivatives, and mixtures thereof;
  - 15 at least one abrasive material selected from oxides, carbonates, quartzes, siliceous chalk, diatomaceous earth, colloidal silicon dioxide, alkali metasilicates, organic abrasive materials selected from polyolefins, polyethylenes, polypropylenes, polyesters, polystyrenes, acetonitrile-butadiene-styrene resins, melamines, polycarbonates, phenolic resins, epoxies and polyurethanes, natural materials selected from rice hulls, corn cobs, and the like, nepheline syenite, or talc and mixtures thereof;
  - 20 water; and
  - optionally, one or more materials selected from perfumes and perfume stabilizers, builders, rheology stabilizers; pH and buffering agents, electrolytes, pigments, colorants and the like.
2. The composition according to claim 1 wherein the surfactant is an amine oxide.
3. The composition according to any one of claims 1 to 2 wherein the surfactant is a mixture of anionic surfactant and amine oxide.
- 25 4. The composition according to any one of claims 1 to 3 wherein the anionic surfactant is selected from alkyl sulfates and alkyl ether sulfates.
5. The composition according to any one of claims 1 to 4 wherein the anionic surfactant is selected from alkyl ether sulfates.
- 30 6. The composition according to any one of claims 1 to 5 wherein the thickener is a mixture of polycarboxylates and clays.

7. The composition according to any one of claims 1 to 5 wherein the thickener is a mixture of polycarboxylates and polyacrylamides.
- 5 8. The composition according to any one of claims 1 to 5 wherein the thickener is a polycarboxylates.
9. The composition according to any one of claims 1 to 8 which contains a rheology modifier.
- 10 10. The composition according to any one of claims 1 to 9 wherein the pH of the composition is 12 or greater.
11. The compositions of the present invention substantially described with reference to the Examples.

## INTERNATIONAL SEARCH REPORT

Internat	Application No
PCT/GB 03/04433	

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7	C11D17/00	C11D3/395	C11D1/83	C11D3/14	C11D3/00
	C11D3/37	C11D3/22	C11D3/12		

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C11D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 96/35771 A (RECKITT & COLMAN INC) 14 November 1996 (1996-11-14) page 4, lines 13-26; claims; examples page 6, line 25 - page 7, line 28 -----	1-11
X	US 4 695 394 A (CHOY CLEMENT K ET AL) 22 September 1987 (1987-09-22) column 13, lines 54-58 - page 14; examples -----	1-11
X	EP 0 649 898 A (CLOROX CO) 26 April 1995 (1995-04-26) page 4, lines 18-50; examples page 8, lines 16-50 ----- -/-	1-11

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

## \* Special categories of cited documents :

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the International filing date
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- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the International filing date but later than the priority date claimed

\*T\* later document published after the International filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

\*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

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\*&\* document member of the same patent family

Date of the actual completion of the International search

18 February 2004

Date of mailing of the International search report

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## INTERNATIONAL SEARCH REPORT

Internati Application No  
PCT/GB 03/04433

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

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X	WO 95/08619 A (CLOROX CO) 30 March 1995 (1995-03-30) page 6, line 12, paragraph 8 - page 9, line 6; claims; examples page 14, lines 17-31 -----	1-11
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